

KS3 Unit Overview – Big Picture

Subject/Year group/Unit Title	Big picture questions	Pupils will focus particularly on the following statements from the programme of study:
<p style="text-align: center;">Chemistry Year 8C Chemical Patterns</p> <p>Lesson 1: Discovering Elements Lesson 2: Elements and symbols Lesson 3: Development of the Periodic Table Lesson 4: Periodic Table Metals and non-metals Lesson 5: Periodic Table Group 1 Lesson 6: Periodic Table Group 7 and 0 Lesson 7: Compounds, Elements and Formulae Lesson 8: Compounds, Elements and Formulae (2) Lesson 9: Recognising Chemical Change Lesson 10 : Types of Chemical Change Lesson 11: Heat in chemical reactions Lesson 12: Assessment Lesson Lesson 13: EOU Assessment</p>	<ul style="list-style-type: none"> • Development of the periodic table • Understanding the periodic table • Periodic trends • Chemical and physical changes • Writing word and simple symbol equations 	<p>CPe1: the varying physical and chemical properties of different elements CPe2: the principles underpinning the Mendeleev Periodic Table CPe3: the Periodic Table: periods and groups; metals and non-metals CPe4: how patterns in reactions can be predicted with reference to the Periodic Table PMPH5: the difference between chemical and physical changes* CCh1: chemical reactions as the rearrangement of atoms CCh2: representing chemical reactions using formulae and using equations</p>
<p>Assessment tasks</p>	<p>As FCJ educators, we will focus on the FCJ values by:</p>	<p>We will ensure students skills in reading, writing, communication and mathematics are enhanced by:</p>
<ul style="list-style-type: none"> • Homework • Formative Badger Assessment • Summative end of unit test 	<p>Excellence – set highest possible standards for all learners Companionship – teamwork when completing practical investigations, respect during class discussions</p>	<p>Mathematics – graph skills, time line, balancing equations, groups and periods Reading – within lessons themselves and literacy news reports Writing – extended Badger assessment Communication – discussions within lessons,</p>

	<p>Dignity – class discussions and Q&A, ensuring everyone is listened to and their views heard</p> <p>Justice - class discussions and Q&A, ensuring everyone is listened to and their views heard</p> <p>Hope – highlight progress in science and innovation to inspire learners</p> <p>Gentleness – classroom management in a firm but fair and gentle manner</p>	
We are supporting progression from KS2 in this unit by:	We are supporting progression to KS4 in this unit by:	Misconceptions and how they will be addressed
Learners have an understanding of physical and chemical changes. They have been introduced to the particle theory.	<p>Development of ideas for periodic table. The arrangement of atoms in the periodic table.</p> <p>Groups and periods. Properties of elements.</p> <p>Periodic trends. Representing substances using formulae. Representing chemical changes using equations.</p>	Difference between chemical and physical changes. Atoms in formulae. Balancing equations.