

Bellerive FCJ Catholic College



Department: Chemistry A Level

Year Group: 13

Term	Learning Focus	Key Knowledge and Skills	Assessment	Challenge and Enrichment
1	<p>Physical Chemistry 3.1.8 Thermodynamics – Started in Y12 3.1.9 Rate equations 3.1.10 Equilibrium constant K_p for homogeneous systems 3.1.12 Acids and bases</p> <p>Organic Chemistry 3.3.9 Carboxylic acids and derivatives 3.3.10 Aromatic chemistry 3.3.11 Amines 3.3.12 Polymers 3.3.13 Amino acids, proteins, and DNA 3.3.14 Organic synthesis 3.3.15 Nuclear magnetic resonance spectroscopy 3.3.16 Chromatography</p>	<ul style="list-style-type: none"> Gibbs free-energy change ΔG and entropy change ΔS Determination of rate equation Equilibrium constant K_p for homogeneous systems Brønsted-Lowry acid-base equilibria in aqueous solutions, Definition and determination of pH, The ionic product of water K_w, Weak acids and bases; K_a for weak acids Acylation Bonding, Electrophilic substitution Preparation, Base Properties, Nucleophilic properties Condensation polymers, Biodegradability and disposal of polymers, Amino acids, Proteins, Enzymes, DNA, Action of anti-cancer drugs Organic synthesis 	<p>Y13 mock exams Thermodynamics HW</p> <p>Required Practical 7b Required Practical 7a Rate equations HW K_p HW</p> <p>Required Practical 10a Required Practical 10b Carboxylic acids and derivatives HW Amines HW</p>	<p>Allery Chemistry (You Tube) A Level Chemistry (Resources) Doc Brown</p>
2	<p>Physical Chemistry 3.1.12 Acids and bases 3.1.11 Electrode potential and electrochemical cells</p> <p>Organic Chemistry</p>	<ul style="list-style-type: none"> pH curves, titrations and indicators, Buffer action Electrode potentials and cells, Commercial applications of electrochemical cells 	<p>Required practical 9 Acids and bases HW Required practical 8 Electrode potentials HW</p>	

	<p>3.3.15 Nuclear magnetic resonance spectroscopy 3.3.16 Chromatography</p> <p>Inorganic Chemistry 3.2.4 Properties of Period 3 elements and their oxides 3.2.5 Transition metals</p> <p>3.2.6 Reactions of ions in aqueous solution</p>	<p>Nuclear magnetic resonance spectroscopy</p> <ul style="list-style-type: none"> • Chromatography • Properties of Period 3 elements and their oxides • General Properties of the Transition Metals, Substitution reactions, Shapes of complex ions, Formation of coloured ions, Variable oxidation, Catalysts • Reactions of ions in aqueous solution 	<p>NMR HW</p> <p>Required Practical 12 Chromatography HW</p> <p>Required practical 11 Transition metals HW Reactions of ions in aqueous solutions HW</p>	
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