

Bellerive FCJ Catholic College



Department: Chemistry A Level

Year Group: 12

Term	Learning Focus	Key Knowledge and Skills	Assessment	Challenge and Enrichment
1	Physical Chemistry 3.1.1 Atomic structure 3.1.2 Amount of substance 3.1.3 Bonding 3.1.4 Energetics Organic Chemistry 3.3.1 Introduction to organic chemistry 3.3.2 Alkanes 3.3.3 Halogenoalkanes	<ul style="list-style-type: none"> Fundamental particles, Mass number and isotopes, Electron configuration Relative atomic mass and relative molecular mass, The mole and Avogadro constant, The ideal gas equation, Empirical and molecular formula, Balanced equations and associated calculations Ionic bonding, Nature of covalent and dative covalent bonds, Metallic bonding, Bonding and physical properties, Shapes of simple molecules and ions, Bond Polarity, Forces between molecules Enthalpy change, Calorimetry Nomenclature, Reaction mechanisms, Isomerism, Fractional distillation of crude oil, Modification of alkanes by cracking, Combustion of alkanes, Chlorination of alkanes Nucleophilic substitution, Elimination 	Atomic structure HW + test Required Practical 1a and 1b Amount of substance HW + test Bonding HW + test Intro to Organic HW Alkanes HW	Allery Chemistry (You Tube) A Level Chemistry (Resources) Doc Brown
2	Physical Chemistry 3.1.4 Energetics 3.1.5 Kinetics 3.1.7 Oxidation, reduction and redox Organic Chemistry 3.3.3 Halogenoalkanes	<ul style="list-style-type: none"> Calorimetry, Applications of Hess's law, Bond enthalpies Collision theory, Maxwell–Boltzmann distribution, Effect of concentration and pressure, Effect of catalysts Redox Ozone depletion 	Y12 mock examinations Required Practical 2 Energetics HW Required Practical 6 Kinetics HW Physical Test Halogenoalkanes HW	

	<p>3.3.4 Alkenes</p> <p>3.3.5 Alcohols</p> <p>3.3.6 Organic Analysis</p>	<ul style="list-style-type: none"> Structure, bonding and reactivity, Addition reactions of alkenes, Addition reactions of alkenes, Addition polymers Oxidation of alcohols, Elimination Identification of functional groups by test-tube reactions, Mass Spectrometry, Infra-red spectroscopy 	<p>Alkenes HW</p> <p>Required Practical 5a</p> <p>Required Practical 5b</p> <p>Alcohols HW</p> <p>Organic Analysis HW</p>	
3	<p>Physical Chemistry</p> <p>3.1.6 Chemical equilibria</p> <p>3.1.7 Oxidation, reduction and redox</p> <p>Inorganic Chemistry</p> <p>3.2.1 Periodicity</p> <p>3.2.2 Group II</p> <p>3.2.3 Group VII</p> <p>Physical Chemistry</p> <p>Year 2 Topics:</p> <p>3.1.8 Thermodynamics</p> <p>Organic Chemistry</p> <p>Year 2 Topics:</p> <p>3.3.7 Optical Isomerism</p> <p>3.3.8 Aldehydes and ketones</p> <p>3.3.9 Carboxylic acids and derivatives</p>	<ul style="list-style-type: none"> Chemical equilibria and Le Chatelier's principle, Equilibrium constant K_c for homogeneous systems Classification, Physical properties of the Period 3 elements Group 2, the alkaline earth metals Trends in properties (Group VII), Uses of chlorine and chlorate(I) Born-Haber Cycles, Gibbs free-energy change ΔG and entropy change ΔS Optical isomerism Aldehydes and ketones Carboxylic acids and esters, Acylation 	<p>Required Practical 3</p> <p>Organic test</p> <p>Y12 mock examinations</p> <p>Redox HW</p> <p>Equilibria HW</p> <p>Periodicity HW</p> <p>Group II HW</p> <p>Group VII HW</p> <p>Thermodynamics HW</p> <p>Isomerism HW</p> <p>Aldehydes and ketones HW</p>	